

# Coupled Thermal-Hydrodynamic-Mechanical -Chemical (THMC) Processes in CO<sub>2</sub> Geological Storage 地质封存的热-流-力-化耦合过程

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# **Outlines**

- **D** Background
- □ Interactions between THMC processes
- **D** Individual processes
- **Coupled processes**
- **Conclusions and recommendations**





# Background

### CO<sub>2</sub> Geological utilization and storage (CGUS)



•EOR驱油
•ECBM驱煤层气
•EGR驱天然气
•ESGR驱页岩气
•EGS强化采热
•EUL溶浸采铀
•EWR驱水

#### Common scientific issue? 共同的基础科学问题



# Background



- Large scale: tens of millions ton of CO<sub>2</sub> injected at single site, area of about 1000km<sup>2</sup> involved
- Long term: hundreds of years
- Multi-physical coupling: T-H-M-C coupling

Aims: Current understanding, theory framework

![](_page_4_Picture_0.jpeg)

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![](_page_4_Picture_7.jpeg)

![](_page_5_Picture_0.jpeg)

![](_page_5_Figure_1.jpeg)

![](_page_6_Picture_0.jpeg)

# **Outlines**

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![](_page_6_Picture_7.jpeg)

![](_page_7_Picture_0.jpeg)

# Individual process-Hydrodynamic process

# Multiphase and multicomponent

CO<sub>2</sub>, H<sub>2</sub>O, NaCl Aqueous, gaseous

![](_page_7_Picture_4.jpeg)

#### Darcy's law for Multiphase flow

#### Mass conservation

$$\frac{d}{dt}\int_{V_n} M^{\kappa} dV = \int_{\Gamma_n} \mathbf{F}^{\kappa} \bullet \mathbf{n} d\Gamma + \int_{V_n} q^{\kappa} dV$$

$$M^{\kappa} = \sum_{\beta=A,G} \phi S_{\beta} \rho_{\beta} X_{\beta}^{\kappa}, \quad \kappa = w, i, g$$

$$\mathbf{F}_{\beta}^{\kappa} = -\frac{k \frac{k_{r\beta} \rho_{\beta}}{\mu_{\beta}} X_{\beta}^{\kappa} (\nabla P_{\beta} - \rho_{\beta} \mathbf{g}) + \mathbf{J}_{\beta}^{\kappa}, \quad \kappa = w, i, g$$

$$\sum_{\beta=A.G} S_{\beta} = 1 \qquad P_g - P_l = P_c$$

![](_page_7_Picture_11.jpeg)

![](_page_8_Picture_0.jpeg)

# Single process-Hydrodynamic process

#### **Key Parameter-permeability**

![](_page_8_Figure_3.jpeg)

![](_page_8_Picture_4.jpeg)

![](_page_9_Picture_0.jpeg)

# Single process-Hydrodynamic process

#### **Key Parameter-relative permeability**

![](_page_9_Figure_3.jpeg)

![](_page_9_Figure_4.jpeg)

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![](_page_10_Picture_0.jpeg)

## Single process-Hydrodynamic process

### **Fluid property**

#### 473-Density (kg/m3) Temperature (K) 423-Committee (1967) 373-323-**Mixtures** Density, viscosity, 473enthalpy Viscosity (\*10-5 Pa s) Temperature (K) 423-373-323critical saturation point line Pressure (bar) 273-473-60 · Enthalpy (kJ/kg) Temperature (K) 423-373-323-Temperature (°C)

Pure CO<sub>2</sub>

Pressure (bar)

#### Pure H<sub>2</sub>O

International Formulation

4 開 出 以 石

![](_page_11_Picture_0.jpeg)

# **Coupled processes-Hydrodynamic process**

#### MRI images of CO2 displacement (Jiang LL, 2016)

![](_page_11_Picture_3.jpeg)

**Pore-scale** 

![](_page_11_Picture_5.jpeg)

#### Numerical simulation of CO<sub>2</sub> injection (site-scale)

![](_page_11_Figure_7.jpeg)

![](_page_12_Picture_0.jpeg)

# Coupled processes-Hydrodynamic process

#### Impure CO<sub>2</sub> injection at Tongliao

![](_page_12_Figure_3.jpeg)

Lei, et al. 2016

![](_page_12_Figure_5.jpeg)

![](_page_13_Picture_0.jpeg)

#### **Coupled processes-Hydrodynamic process** Monitoring of CO<sub>2</sub> migration Boait, et al. 2012

2004

![](_page_13_Picture_2.jpeg)

![](_page_13_Picture_3.jpeg)

![](_page_13_Picture_4.jpeg)

#### **Sleipner Field**

2006

![](_page_13_Picture_7.jpeg)

![](_page_13_Picture_8.jpeg)

![](_page_13_Figure_9.jpeg)

#### Shenghua CCS site

Li, et al. 2016

![](_page_13_Figure_12.jpeg)

![](_page_14_Picture_0.jpeg)

# Single process-Thermal process

**Must be coupled** 

#### **Energy conservation**

with H!  $\frac{d}{dt} \int_{V_n} M^{\kappa} dV = \int_{\Gamma_n} \mathbf{F}^{\kappa} \bullet \mathbf{n} d\Gamma + \int_{V_n} q^{\kappa} dV$ convection conduction  $\mathbf{F}_{\beta}^{\kappa+1} = -\frac{\lambda \nabla T}{\Gamma} + \sum_{\beta} h_{\beta} F_{\beta}$  $M^{\kappa+1} = (1-\phi)\rho_R C_R T + \sum_{\beta \neq \alpha} \phi S_{\beta} \rho_{\beta} u_{\beta}$  $\beta = A, G$ 

![](_page_15_Picture_0.jpeg)

## Single process-Mechanical process

#### **Motion equation**

$$-G\nabla^{2}w_{x} - \frac{G}{1-2\upsilon}\frac{\partial}{\partial x}\left(\frac{\partial w_{x}}{\partial x} + \frac{\partial w_{y}}{\partial y} + \frac{\partial w_{z}}{\partial z}\right) + \frac{\partial P}{\partial x} + 3\beta_{T}K\frac{\partial T}{\partial x} = 0$$
  
$$-G\nabla^{2}w_{y} - \frac{G}{1-2\upsilon}\frac{\partial}{\partial y}\left(\frac{\partial w_{x}}{\partial x} + \frac{\partial w_{y}}{\partial y} + \frac{\partial w_{z}}{\partial z}\right) + \frac{\partial P}{\partial y} + 3\beta_{T}K\frac{\partial T}{\partial y} = 0$$
  
$$-G\nabla^{2}w_{z} - \frac{G}{1-2\upsilon}\frac{\partial}{\partial z}\left(\frac{\partial w_{x}}{\partial x} + \frac{\partial w_{y}}{\partial y} + \frac{\partial w_{z}}{\partial z}\right) + \frac{\partial P}{\partial z} + 3\beta_{T}K\frac{\partial T}{\partial z} = \gamma_{sat}$$

#### Must be coupled with H and T !

#### **Stress-strain**

$$\sigma'_{x} = 2G(\frac{\upsilon}{1-2\upsilon}\varepsilon_{v} + \varepsilon_{x}) + 3\beta_{T}K\Delta T$$

$$\sigma'_{y} = 2G(\frac{\upsilon}{1-2\upsilon}\varepsilon_{v} + \varepsilon_{y}) + 3\beta_{T}K\Delta T$$

$$\sigma'_{z} = 2G(\frac{\upsilon}{1-2\upsilon}\varepsilon_{v} + \varepsilon_{z}) + 3\beta_{T}K\Delta T$$

$$\tau'_{yz} = G\gamma_{yz}, \tau'_{zx} = G\gamma_{zx}, \tau'_{xy} = G\gamma_{xy}$$

#### **Strain-displacement**

$$\varepsilon_{x} = -\frac{\partial w_{x}}{\partial x}, \gamma_{yz} = -\left(\frac{\partial w_{y}}{\partial z} + \frac{\partial w_{z}}{\partial y}\right)$$
$$\varepsilon_{y} = -\frac{\partial w_{y}}{\partial y}, \gamma_{zx} = -\left(\frac{\partial w_{z}}{\partial x} + \frac{\partial w_{x}}{\partial z}\right)$$
$$\varepsilon_{z} = -\frac{\partial w_{z}}{\partial z}, \gamma_{xy} = -\left(\frac{\partial w_{x}}{\partial y} + \frac{\partial w_{y}}{\partial x}\right)$$

![](_page_16_Picture_0.jpeg)

# Single process-Chemical process

![](_page_16_Figure_2.jpeg)

![](_page_17_Picture_0.jpeg)

# Single process-Chemical process

#### **Chemical reaction-gas solution**

![](_page_17_Figure_3.jpeg)

![](_page_18_Picture_0.jpeg)

# Single process-Chemical process

#### **Chemical reaction-water/rock reaction**

![](_page_18_Figure_3.jpeg)

#### **Reaction rate for kinetic minerals**

$$k(T) = k_{25}^{nu} \exp\left[\frac{-E_a^{nu}}{R}\left(\frac{1}{T} - \frac{1}{298.15}\right)\right]$$

$$r_m = \pm k(T)_m \left|1 - \left(\frac{Q_m}{K_m}\right)^{\theta}\right|^{\eta} + k_{25}^{H} \exp\left[\frac{-E_a^{H}}{R}\left(\frac{1}{T} - \frac{1}{298.15}\right)\right]a_H^{n_H}$$

$$+ k_{25}^{OH} \exp\left[\frac{-E_a^{OH}}{R}\left(\frac{1}{T} - \frac{1}{298.15}\right)\right]a_{OH}^{n_{OH}}$$

![](_page_19_Picture_0.jpeg)

### **Coupled processes-**Chemical process CO<sub>2</sub>-water-rock reaction-batch experiment

![](_page_19_Figure_2.jpeg)

#### CO2-water-rock reaction near the wellbore

![](_page_19_Figure_4.jpeg)

![](_page_19_Picture_5.jpeg)

![](_page_20_Picture_0.jpeg)

## **Coupled processes-**Chemical process Long term CO<sub>2</sub>-water-rock reaction

![](_page_20_Picture_2.jpeg)

Kampman, et al. 2014

![](_page_20_Picture_4.jpeg)

![](_page_21_Picture_0.jpeg)

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![](_page_21_Picture_7.jpeg)

![](_page_22_Picture_0.jpeg)

# **Coupled processes-Focus**

Number of process	Coupling
1	Τ、 Η、 <b>C</b> 、 Μ
2	$TH_{v}, HM_{v}, HC_{v}, MC_{v}, TM_{v}, TC$
3	THC、THM、HMC、TCM
4	ТНМС

![](_page_22_Picture_3.jpeg)

![](_page_23_Picture_0.jpeg)

Induced-seismicities can be mitigated by regulations, site selection, evaluation, monitoring and control

![](_page_23_Figure_3.jpeg)

![](_page_23_Picture_4.jpeg)

![](_page_24_Picture_0.jpeg)

stress-strain and pore pressure change 
fault stability
mechanical integrity of the cap-rock
ground surface uplift

**Ground surface uplift** Injection well

![](_page_24_Figure_3.jpeg)

Pressure buildup induced by CO2 injection is the driving force !

![](_page_25_Picture_0.jpeg)

The changes of the stress-strain and pore pressure

![](_page_25_Figure_3.jpeg)

CO<sub>2</sub> injection

- $\rightarrow$  Pore pressure increase
- $\rightarrow$  Effective stress decrease
- $\rightarrow$  Pore volume expansion
- $\rightarrow$  Facilitating the evolution of the

pore pressure

 $\rightarrow$  Promoting CO2 migration

#### Importance:

- Preconditions for analyzing other issues
- Basic understandings for the changes of cap-rock and reservoir

![](_page_26_Picture_0.jpeg)

#### The change of the pore pressure

![](_page_26_Figure_3.jpeg)

Vilarrasa et al. 2010

- Prominent effect of mechanics to pore pressure.
- Notable difference between the aquifer and cap-rock.

![](_page_27_Picture_0.jpeg)

#### Ground surface deformation

![](_page_27_Figure_3.jpeg)

![](_page_28_Picture_0.jpeg)

# Coupled processes-HM processesGround surface deformation $D_{ground}$

![](_page_28_Figure_2.jpeg)

The permeability of the cap-rock has and effect to the surface uplift.
 Temperature and stiffness have a slight effect.

![](_page_28_Figure_4.jpeg)

![](_page_29_Picture_0.jpeg)

![](_page_29_Figure_2.jpeg)

Direct
 response of the subsurface
 mechanical
 issues
 Coinciding
 with the monitoring
 results

In Salah project Left: numerical simulation Right: field monitoring

![](_page_30_Picture_0.jpeg)

ZONE 2

CAPROCK 1

-2000

-1000

DISTANCE FROM INJECTION POINT, X (m)

-1300

-1400 -1500

-3000

# **Coupled processes-HM** processes

![](_page_30_Figure_2.jpeg)

1000

2000

# Mechanical integrity *Elastic model*

Two-dimensional plane-strain model

Partial coupling method

![](_page_30_Picture_6.jpeg)

#### Compressive stress regime $\sigma_r = 1.5 \sigma_r$

• Shear failure is likely to be initiated in shallowly dipping at the interface between the reservoir and cap-rock.

• Low possibility for tensile failure.

![](_page_30_Picture_10.jpeg)

Rutqvist et al. 2007

3000

#### 岩土力学与工程国家重点实验室

#### **Coupled processes-HM processes** anical integrity *Elastic model*

![](_page_31_Figure_2.jpeg)

Extensional stress regime  $\sigma_x = 0.7 \sigma_z$ 

•High potential for shear failure occurs throughout the CO2-storage system and in preferentially steeply dipping fractures.

•High potential for tensile failure occurs in the bottom of the cap-rock

•In compressional stress regime, shear failure is much more likely to occur.

•Compressional stress regime is not favorable for mechanical integrity.

![](_page_32_Picture_0.jpeg)

![](_page_32_Figure_1.jpeg)

![](_page_33_Picture_0.jpeg)

#### Mechanisms for inducing earthquakes

![](_page_33_Figure_3.jpeg)

- The effective stress acting on fault decrease by increasing pore pressure.
- The loading condition on fault has changed.

![](_page_33_Picture_6.jpeg)

![](_page_34_Picture_0.jpeg)

Fault stability- Constitutive model and failure criterion

![](_page_34_Figure_3.jpeg)

![](_page_34_Figure_4.jpeg)

Elastic-perfectly plastic

theory

- Mohr-Coulomb criterion
- Zero cohesive strength.
- The coefficients of the

static friction:  $0.6 \le \mu \le 0.85$ 

![](_page_34_Picture_11.jpeg)

![](_page_34_Picture_12.jpeg)

![](_page_35_Picture_0.jpeg)

#### Stability evaluation of the fault

![](_page_35_Picture_3.jpeg)

Slip tendency on fault surface

The slip tendency is defined as the ratio of the shear stress to the normal effective stress

$$T_s = \frac{\tau}{\sigma_n - p_f}$$

Streit et al.2004

 $\sigma_{\rm n}$  — Normal effective stress

- au Shear effective stress
- $p_f$  Pore pressure

### Evaluating the seismic magnitude

Quantification of the overall size of an earthquake is generally based on the seismic moment  $M_0$  defined for a ruptured patch on a fault by the following:

$$M_o = \mu A d$$
  
 $M = (\log_{10} M_o / 1.5) - 6.1$ 

- *M* Seismic magnitude
- $\mu$  Shear module
- A Reactive area
- d Slip distance

![](_page_35_Picture_17.jpeg)

![](_page_36_Picture_0.jpeg)

![](_page_36_Figure_2.jpeg)

Intention: earthquake magnitude resulting from CO2 injection

 $M_{0} = 4 \times 10^{9} \times 1000 \times 385 \times 0.08 = 1.23 \times 10^{14} Nm$  $M = (\log_{10} M_{o} / 1.5) - 6.1 = 329$ 

![](_page_37_Picture_0.jpeg)

#### Shenhua CCS site

#### temperature

![](_page_37_Figure_4.jpeg)

![](_page_37_Picture_5.jpeg)

![](_page_38_Picture_0.jpeg)

## Summary

![](_page_38_Figure_2.jpeg)

#### 耦合途径:

- 通过物质、能量的交换
- 通过对材料特性、结构、相态的改变

![](_page_38_Picture_6.jpeg)

![](_page_39_Picture_0.jpeg)

1. The key indicators of CGUS project performance are capacity, injectivity, sealing, stability and productivity

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- 2. Prediction is based on detailed geological model and on knowledge and analysis of THMC processes
- 3. Short-term processes in EOR and DSF/EWR have be well understood and reasonably simulated
- 4. Their long-term processes and the other CGUS options need to be studies
- 5. Fully coupled analyses is not always necessary. Identifying the dominated processes is critical

![](_page_39_Figure_6.jpeg)

![](_page_40_Picture_0.jpeg)

# Thank you !

![](_page_40_Picture_2.jpeg)